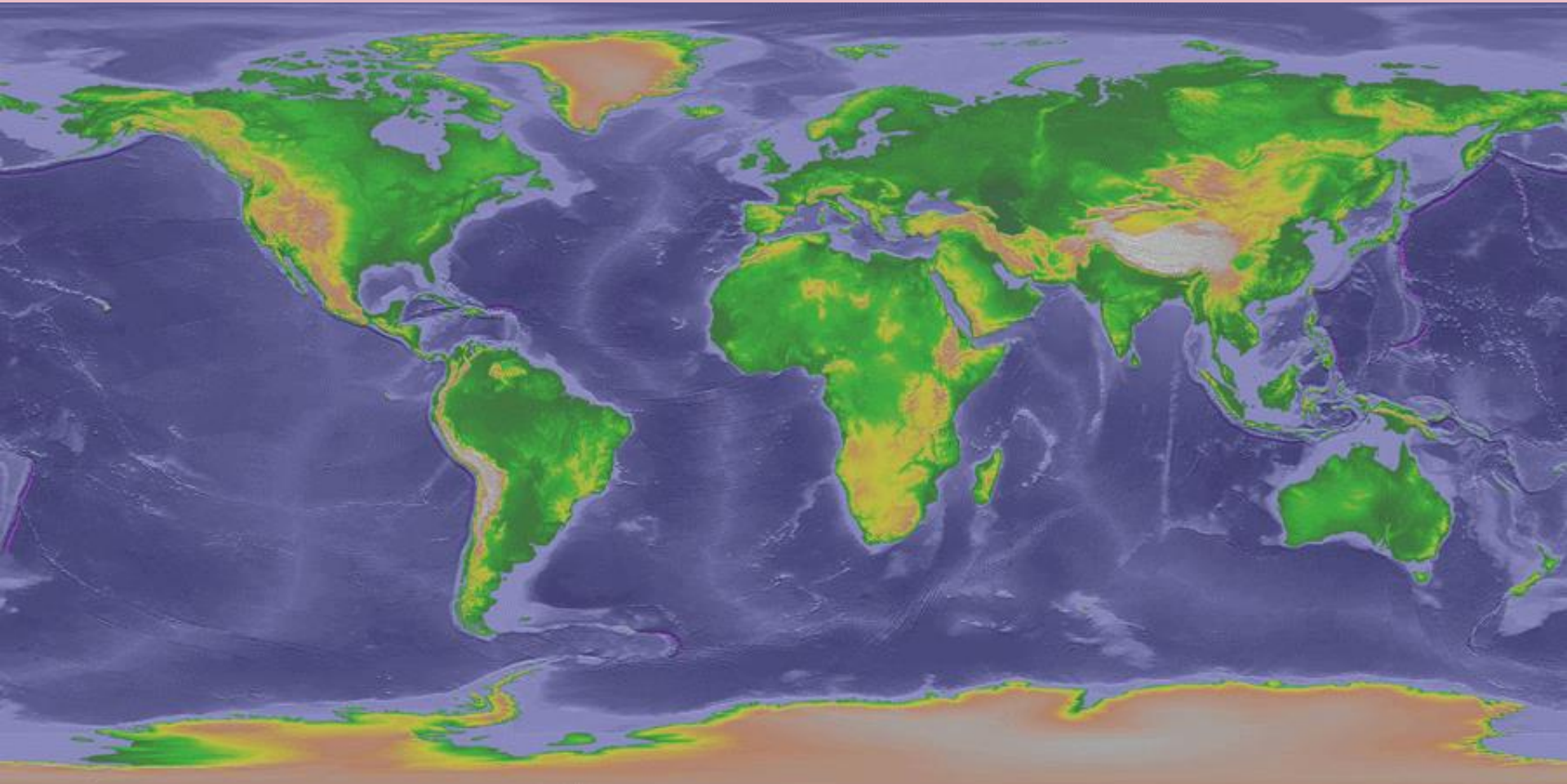


What do you see....



What do you see now....

Less land?!?!?



# Chapter 19

## Global Change

# Global Change

- **Global change-** any chemical, biological or physical property change of the planet.  
Examples: include cold temperatures causing ice ages.
- **Global climate change-** changes in the climate of the Earth.
- **Global warming-** one aspect of climate change, the warming of the oceans, land masses and atmosphere of the Earth.

# The Greenhouse Effect

- When radiation from the sun hits the atmosphere, **1/3 is reflected back.**
- Some of the **UV radiation is absorbed by the ozone** layer and strikes the Earth where it is **converted into low-energy infrared radiation.**
- The infrared radiation then goes back toward the atmosphere where it is **absorbed by greenhouse gasses that radiate most of it back to the Earth.**
- **Greatest contributor is Carbon Dioxide**

**1** Incoming solar radiation consists primarily of UV and visible light.

**2** About one-third of this solar radiation is reflected—from the atmosphere, clouds, and the surface of the planet—back into space.

**4** Much of the emitted infrared radiation from Earth is absorbed by greenhouse gases in the atmosphere. The remainder is emitted into space.

Incoming solar radiation

Reflected by atmosphere and clouds

Reflected from surface

Absorbed by clouds

Absorbed by surface

Outgoing infrared radiation

Greenhouse gases in atmosphere

**3** The remaining solar radiation is absorbed by clouds and the surface of the planet. Both become warmer and then emit infrared radiation.

**5** As the greenhouse gases absorb infrared radiation, they warm and emit infrared radiation, with much of it going back toward Earth. The greater the concentration of greenhouse gases, the more infrared radiation is absorbed and emitted back toward Earth.

On average, Earth is about 60°F, w/o atmosphere it would be 0°F

About 30% of solar energy is reflected back into space, 70% absorbs by Earth (Earth's oceans and land masses eventually radiate heat back out in the form of infrared radiation).

# Greenhouse Gases

-Water vapor

-Nitrous oxide

-Carbon dioxide

-Methane

-Chlorofluorocarbons (CFC's)

*ex. aerosols (ozone depleters, tried to be banned by Montreal Protocol, highest warming potential to carbon dioxide*

**TABLE 19.1**    **The major greenhouse gases**

The major greenhouse gases differ in their ability to absorb infrared radiation and the duration of time that they stay in the atmosphere. The units "ppm" are parts per million.

Greenhouse gas	Concentration in 2010	Global warming potential (over 100 years)	Duration in the atmosphere
Water vapor	Variable with temperature	<1	9 days
Carbon dioxide	390 ppm	1	Highly variable (ranging from years to hundreds of years)
Methane	1.8 ppm	25	12 years
Nitrous oxide	0.3 ppm	300	114 years
Chlorofluorocarbons	0.9 ppm	1,600 to 13,000	55 to >500 years

Source: Data on concentration are from the National Oceanic and Atmospheric Administration. [www.esrl.noaa.gov/gmd/aggi](http://www.esrl.noaa.gov/gmd/aggi). Data on global warming potential are from the United Nations Framework Convention on Climate Change.

Table 19.1

Environmental Science

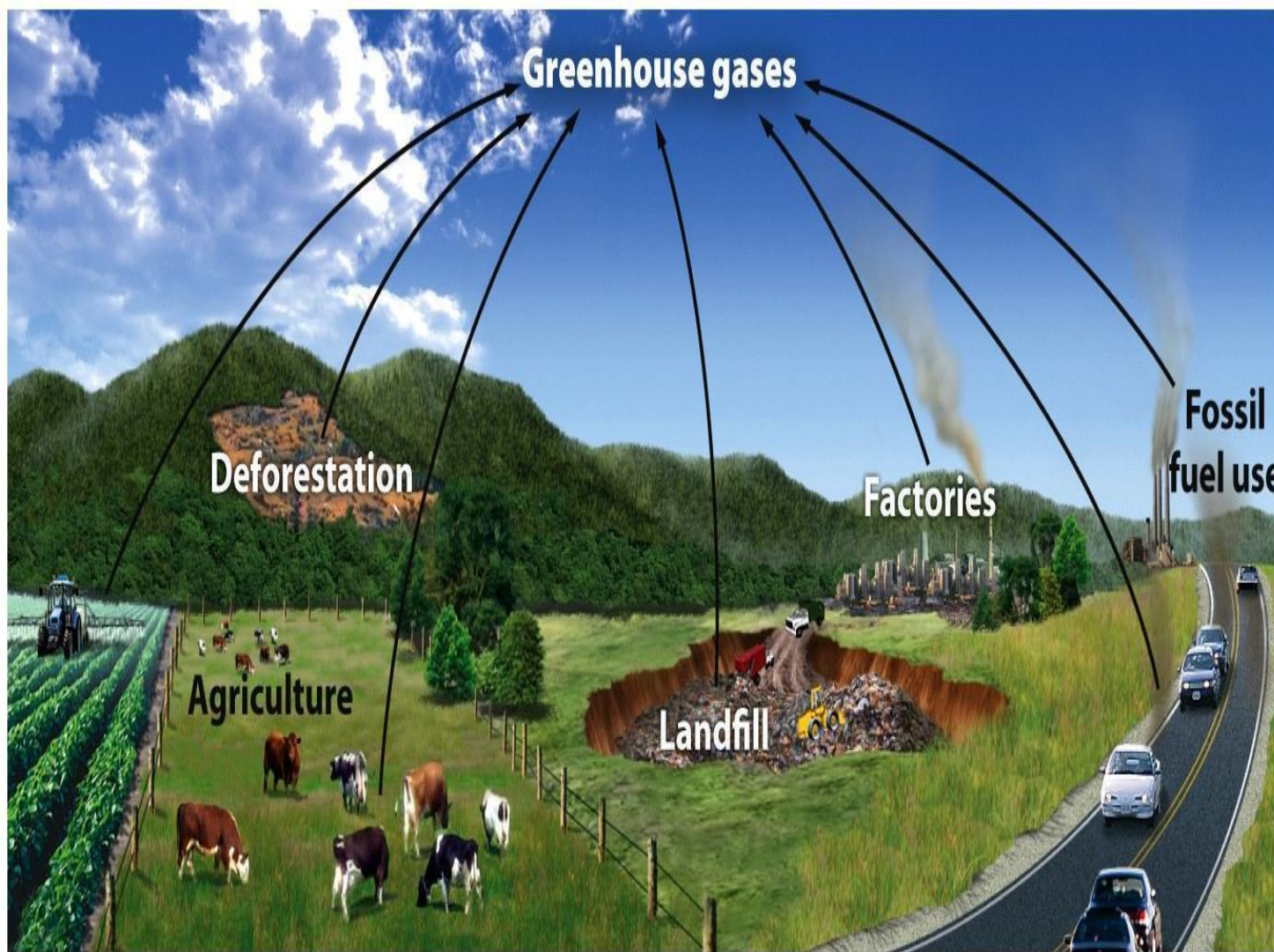
© 2012 W. H. Freeman and Company

# Natural Greenhouse Gases

- **Volcanic eruptions**- mainly carbon dioxide; *causes a cooling of Earth*
- **Methane** – from decomposition
- **Nitrous oxide**- from denitrification
- **Water vapor** – heating of water cycle  
*(greenhouse gas most responsible for trapping the most outgoing infrared radiation)*
- *W/o human activities, greenhouse warming is detrimental to sea life due to the warming and changing of pH levels*



# Anthropogenic Causes of Greenhouse Gases



- Burning of fossil fuels
- Agricultural practices
- Deforestation
- Landfills
- Industrial production- (CFC's are an example)

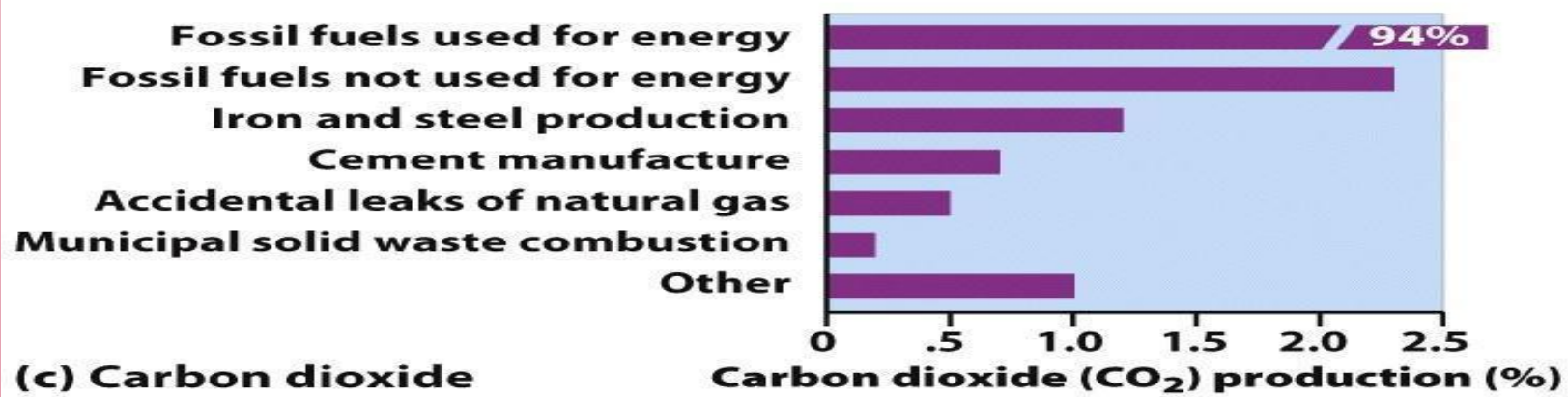
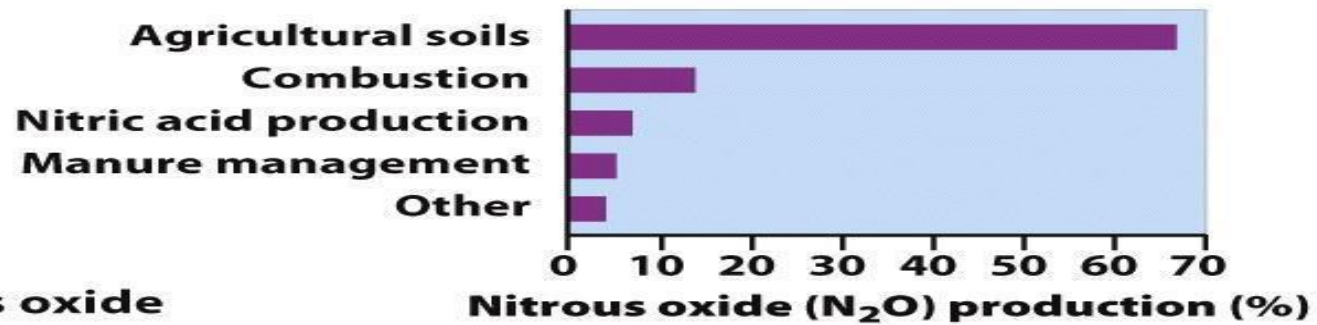
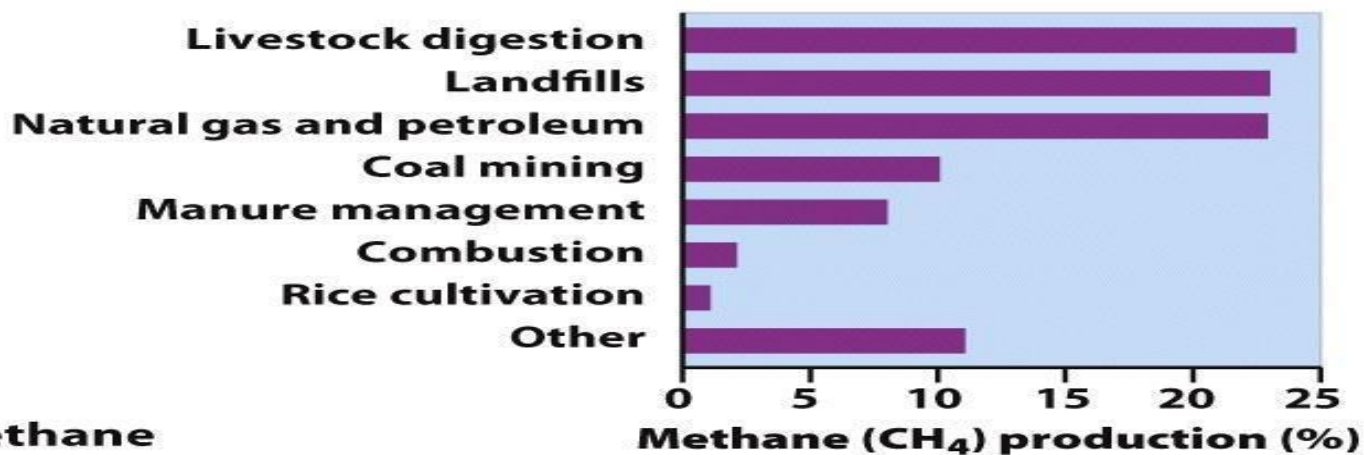
**Figure 19.5**  
*Environmental Science*  
© 2012 W. H. Freeman and Company

# Natural Greenhouse Effect



# Human Enhanced Greenhouse Effect





**Figure 19.6**  
*Environmental Science*  
 © 2012 W. H. Freeman and Company

# Increasing CO<sub>2</sub> Concentrations

- David Keeling began measuring CO<sub>2</sub> in 1958.

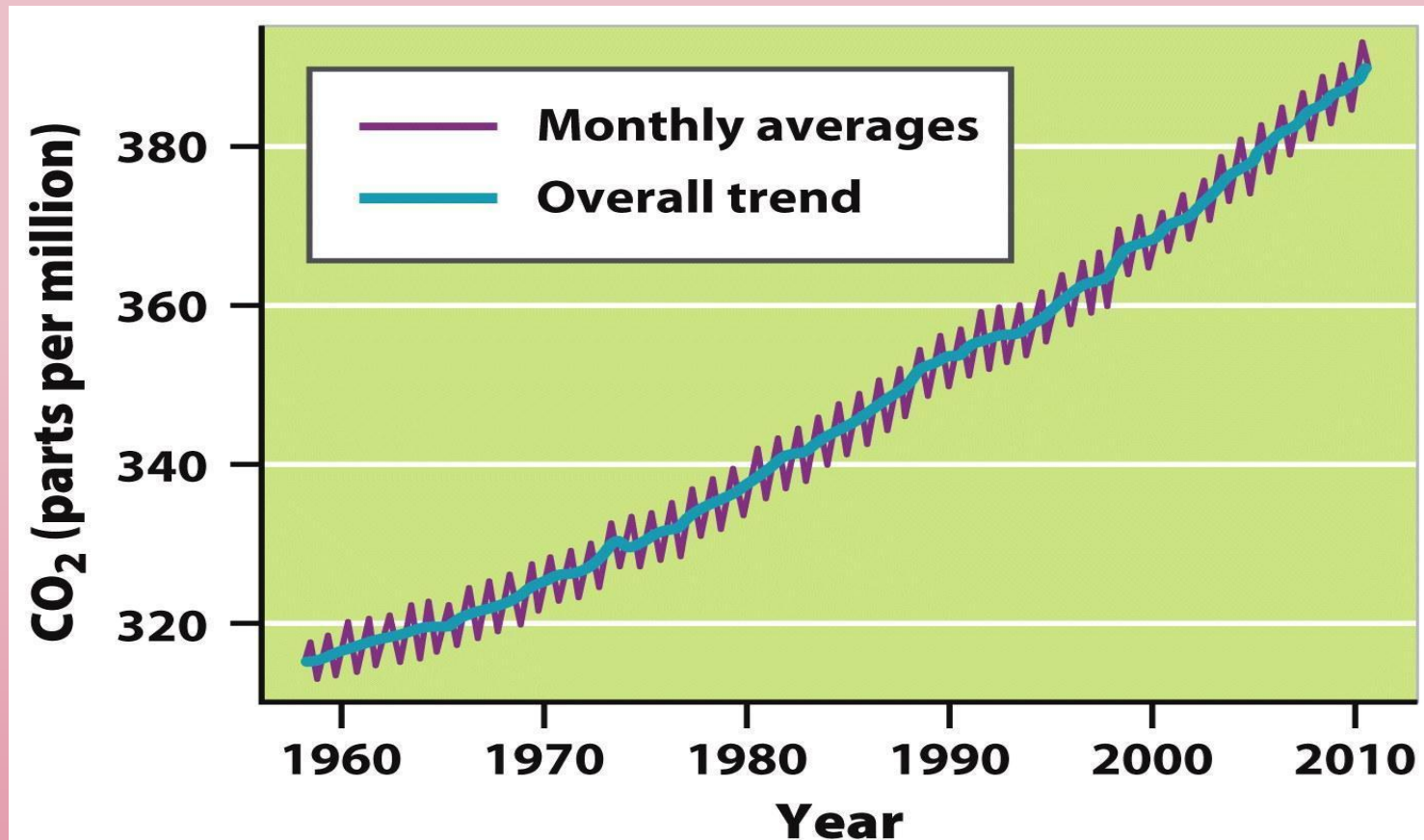
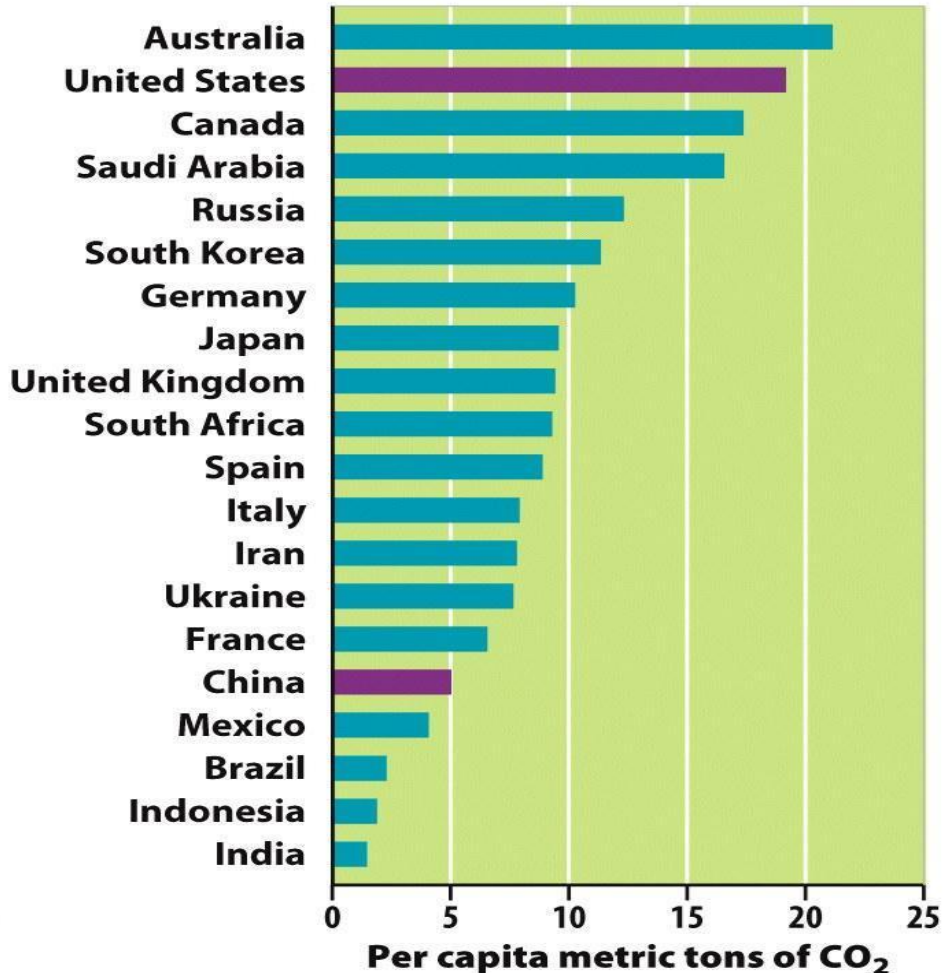
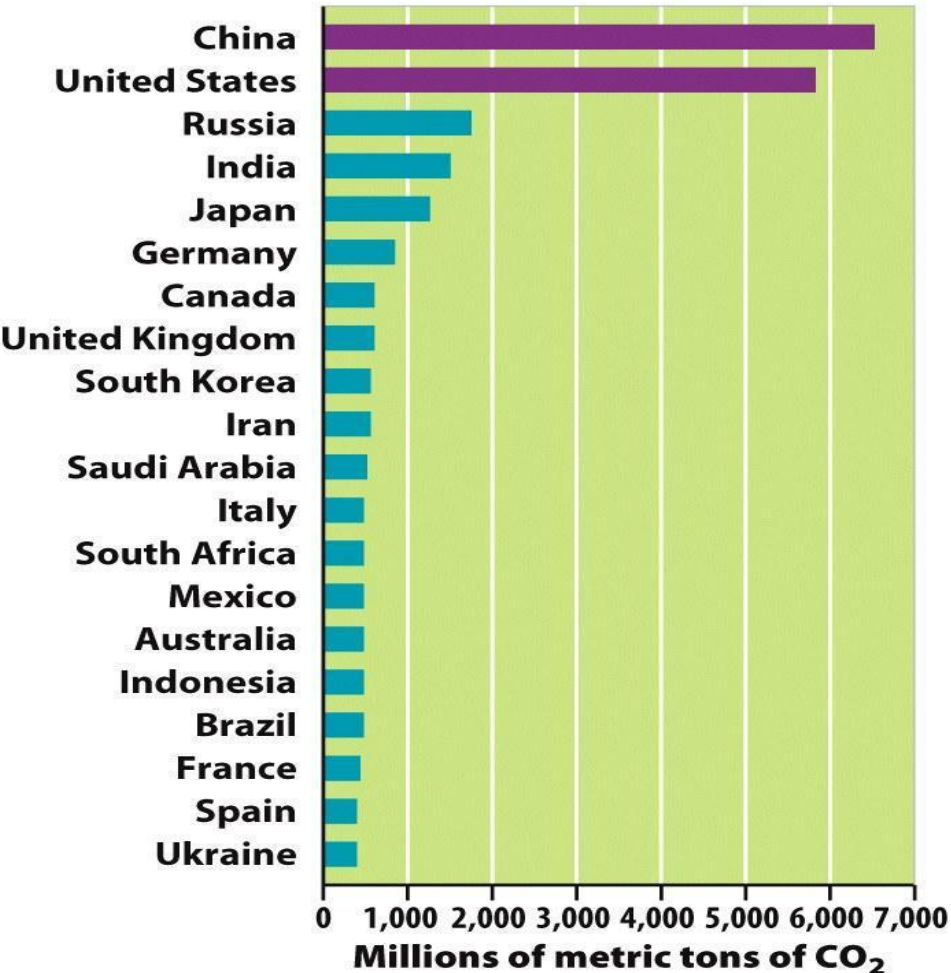


Figure 19.7  
*Environmental Science*  
© 2012 W. H. Freeman and Company

# Emissions from the Developed and Developing World

*(largest contributors are developed and rapidly developing) nations*

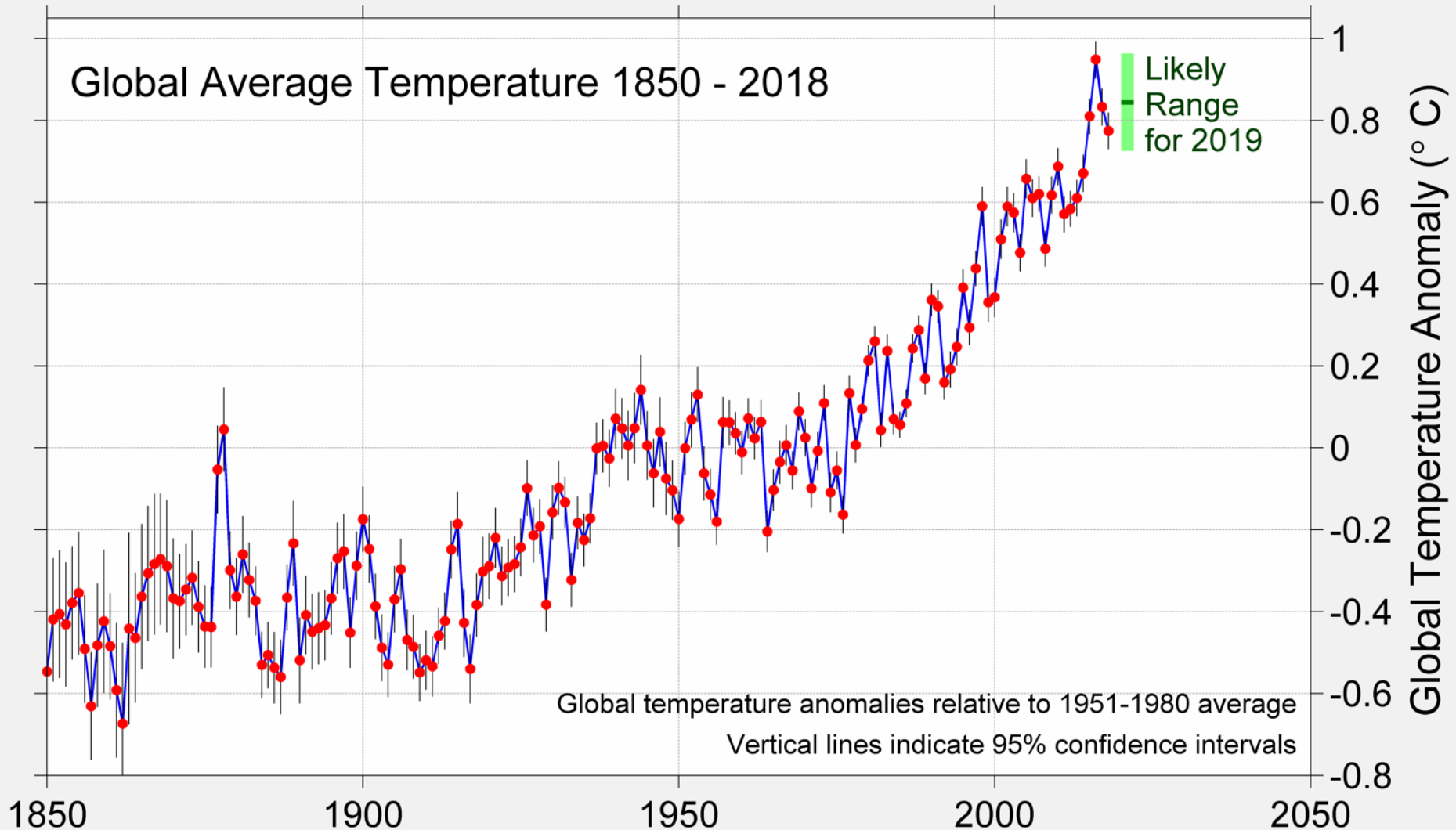


**Total amt of carbon dioxide produced by a country**

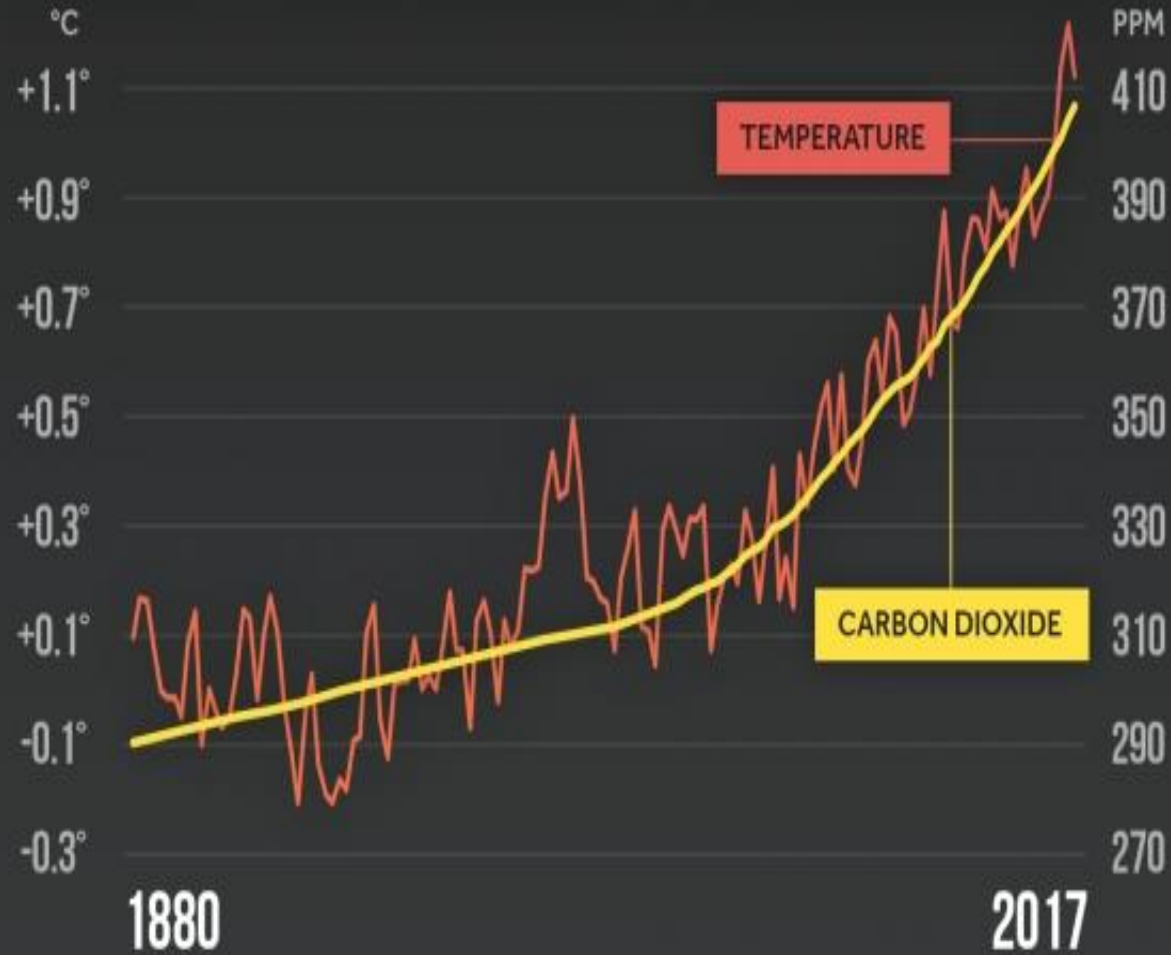
**(Per individual use vs. total population)**

# Global Temperatures since 1850

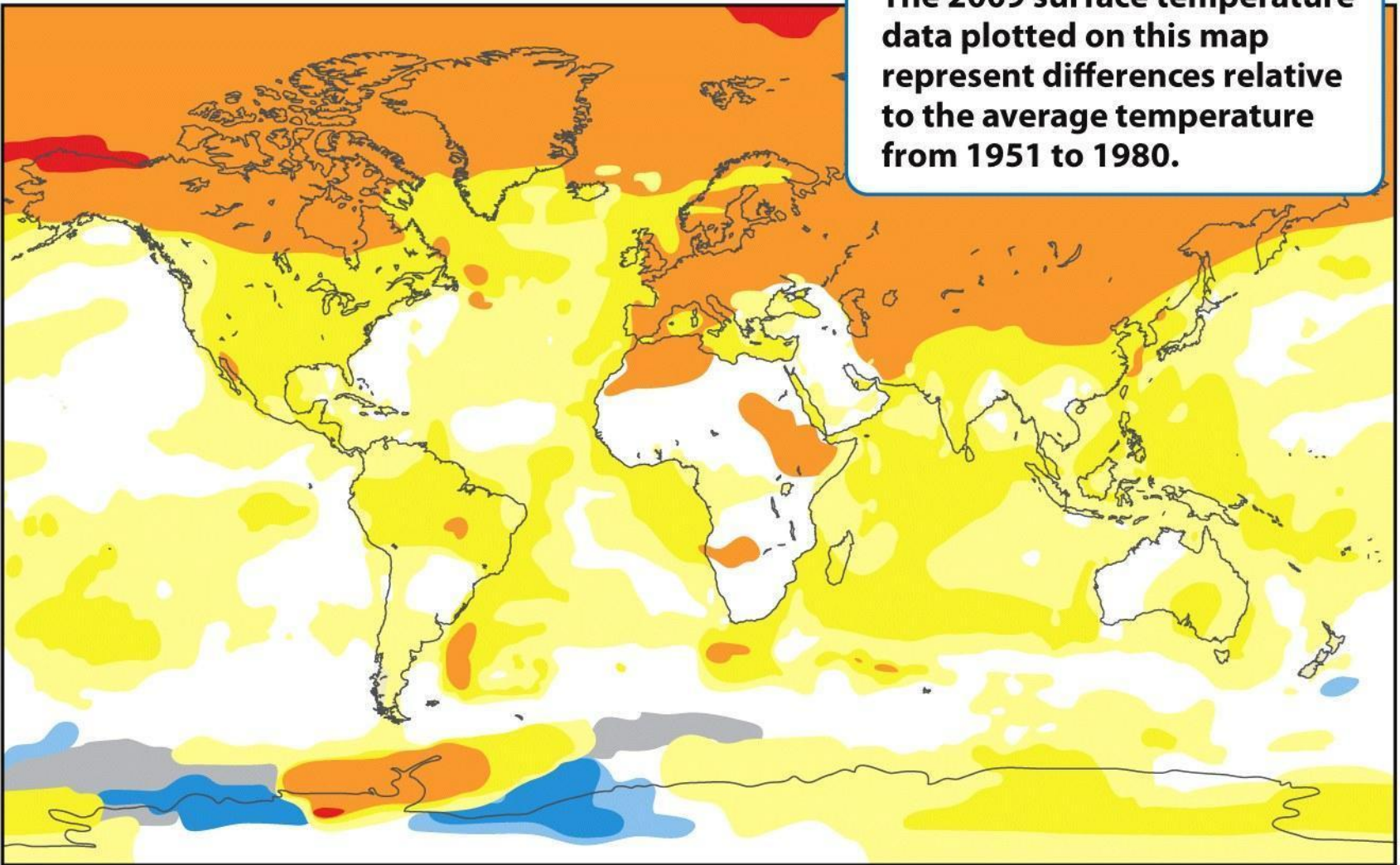
- Since 1880 temperatures have increased  $0.8^{\circ}\text{C}$  ( $1.4^{\circ}\text{F}$ ).



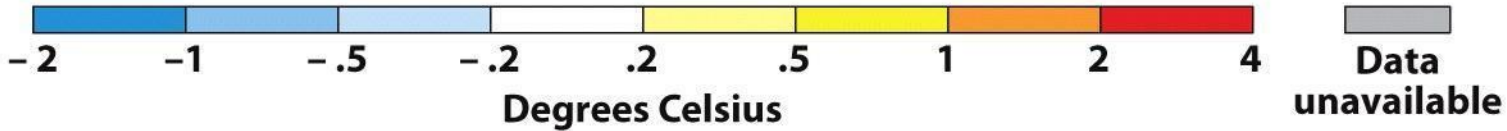
# GLOBAL TEMPERATURE & CARBON DIOXIDE



Global temperature anomalies averaged and adjusted to early industrial baseline (1881-1910)  
Source: NASA GISS, NOAA NCEI, ESRL



The 2009 surface temperature data plotted on this map represent differences relative to the average temperature from 1951 to 1980.



**Figure 19.10**  
*Environmental Science*  
© 2012 W. H. Freeman and Company



# Temperatures and Greenhouse Gas Concentrations in Past 400,000 Years

- No one was around thousands of years ago to measure temperatures so we use **other indirect measurements**. Some of these are
  - **Changes in species compositions** (preserved over millions of years)
  - **Chemical analyses of ice** (formation of *ice cores* based on air bubbles that trap greenhouse gases – measure concentration levels)



Figure 19.12a



Figure 19.12b

# Historic Carbon Dioxide concentrations

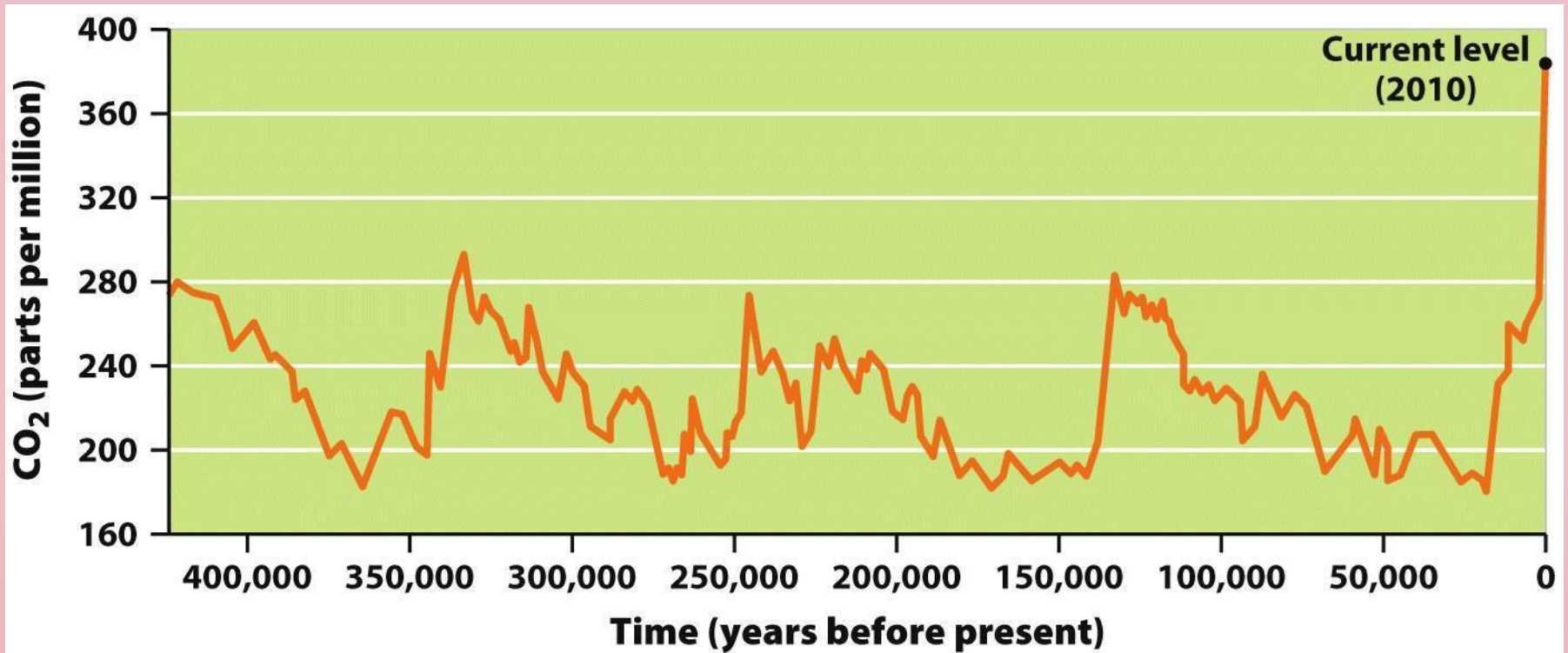
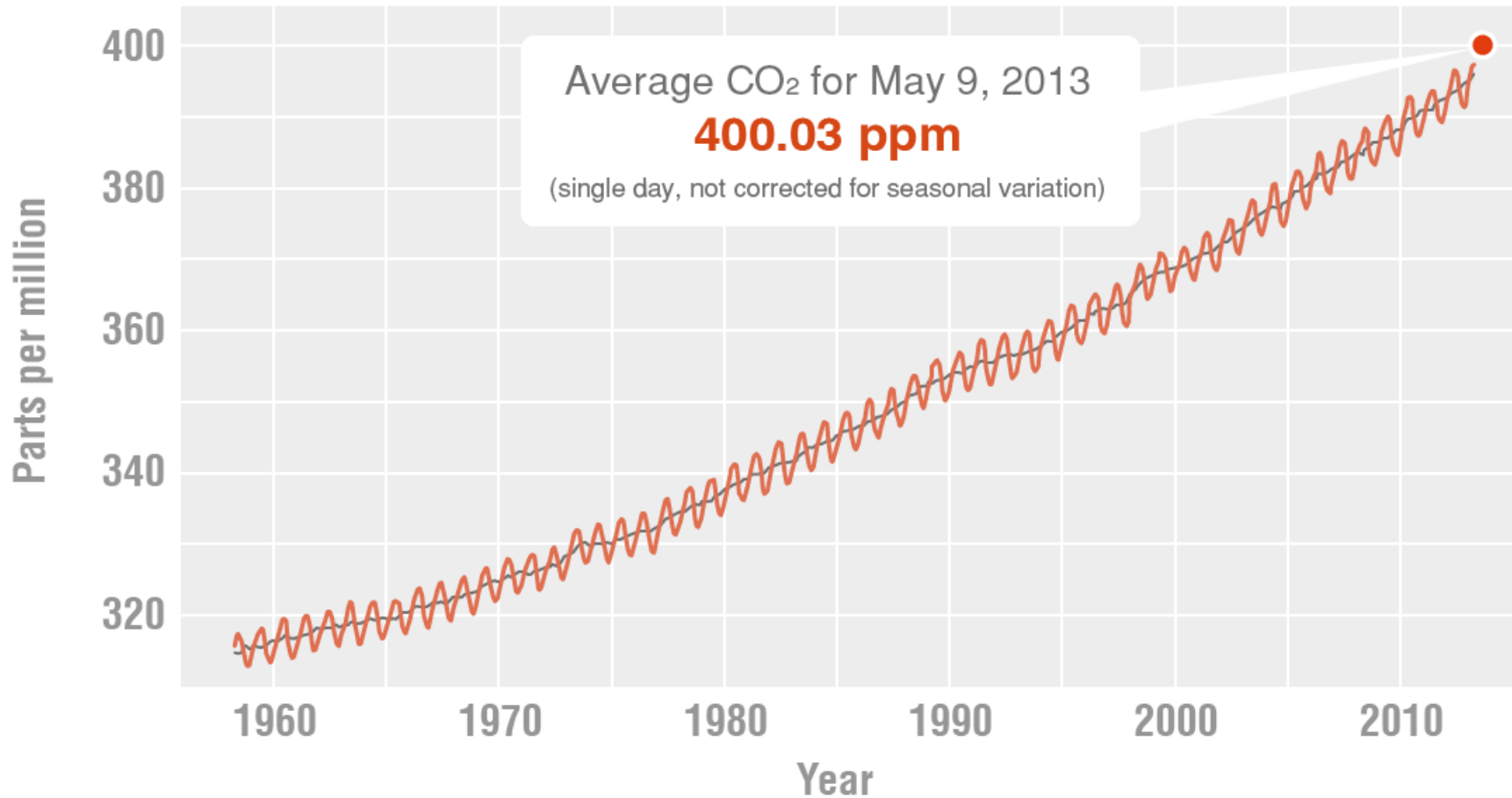


Figure 19.13  
Environmental Science  
© 2012 W. H. Freeman and Company

More than 400,000 years of Carbon Dioxide concentrations never exceeded 300ppm. After 1950, concentrations sharply increased to their current level of 390ppm

# Carbon Dioxide Concentration



Credit: NOAA/Scripps Institution of Oceanography

**Increase use of fossil fuels after the 1950's and still rising**

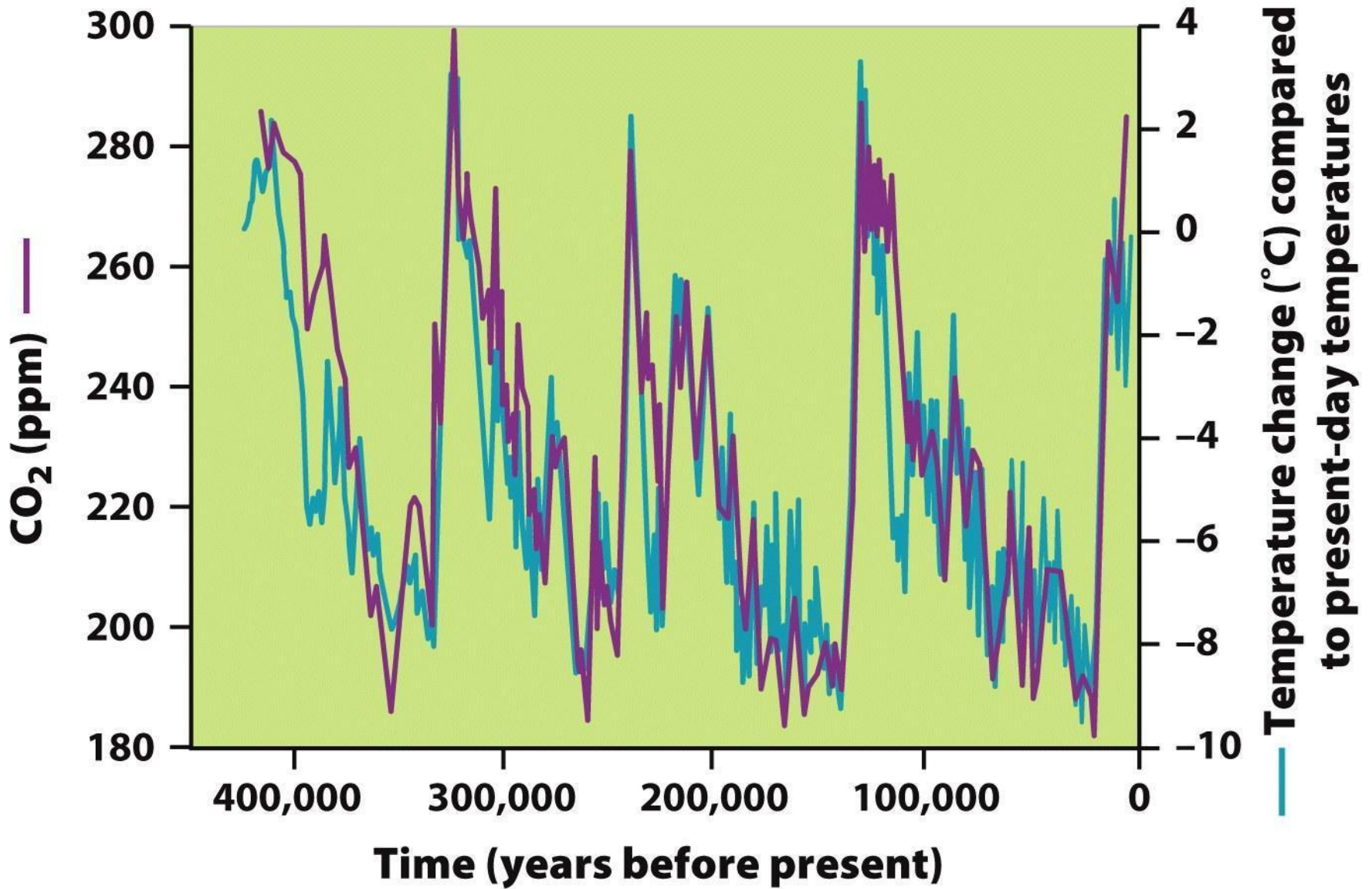
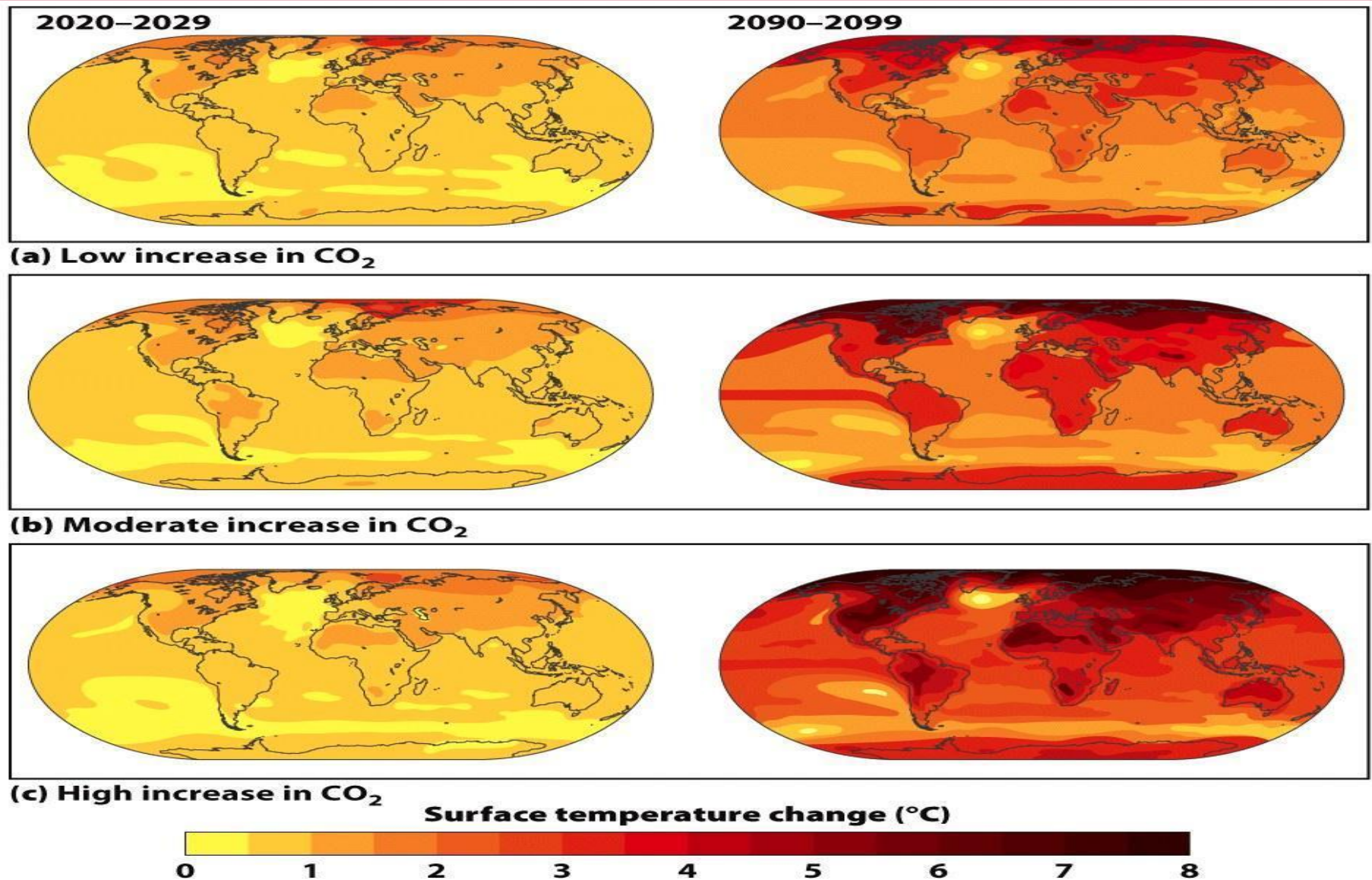


Figure 19.15

Environmental Science

© 2012 W. H. Freeman and Company

# Predictions of increase in global temps by 2100



**Figure 19.17**

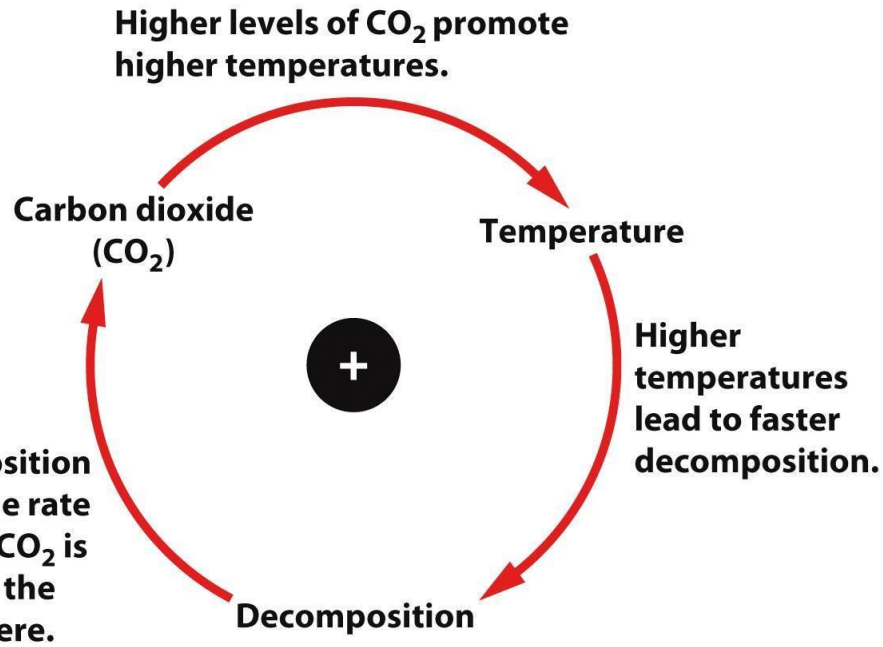
*Environmental Science*

© 2012 W. H. Freeman and Company

# Putting It Together

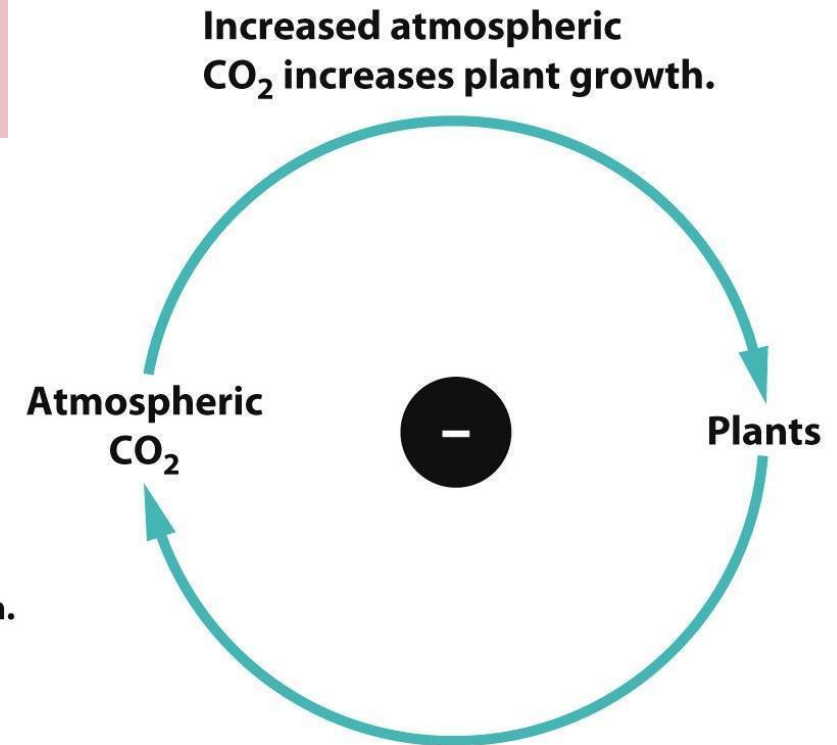
- We know that an increase in CO<sub>2</sub> in the atmosphere causes a greater capacity for warming through the greenhouse effect.
- When the Earth experiences higher temperatures, the oceans warm and cannot contain as much CO<sub>2</sub> gas and, as a result, they release CO<sub>2</sub> into the atmosphere.

# Feedbacks



## Positive feedback system

Figure 19.18a  
Environmental Science  
© 2012 W. H. Freeman and Company



Increased plant growth increases uptake of CO<sub>2</sub> from the atmosphere, thereby decreasing the amount of CO<sub>2</sub> in the atmosphere.

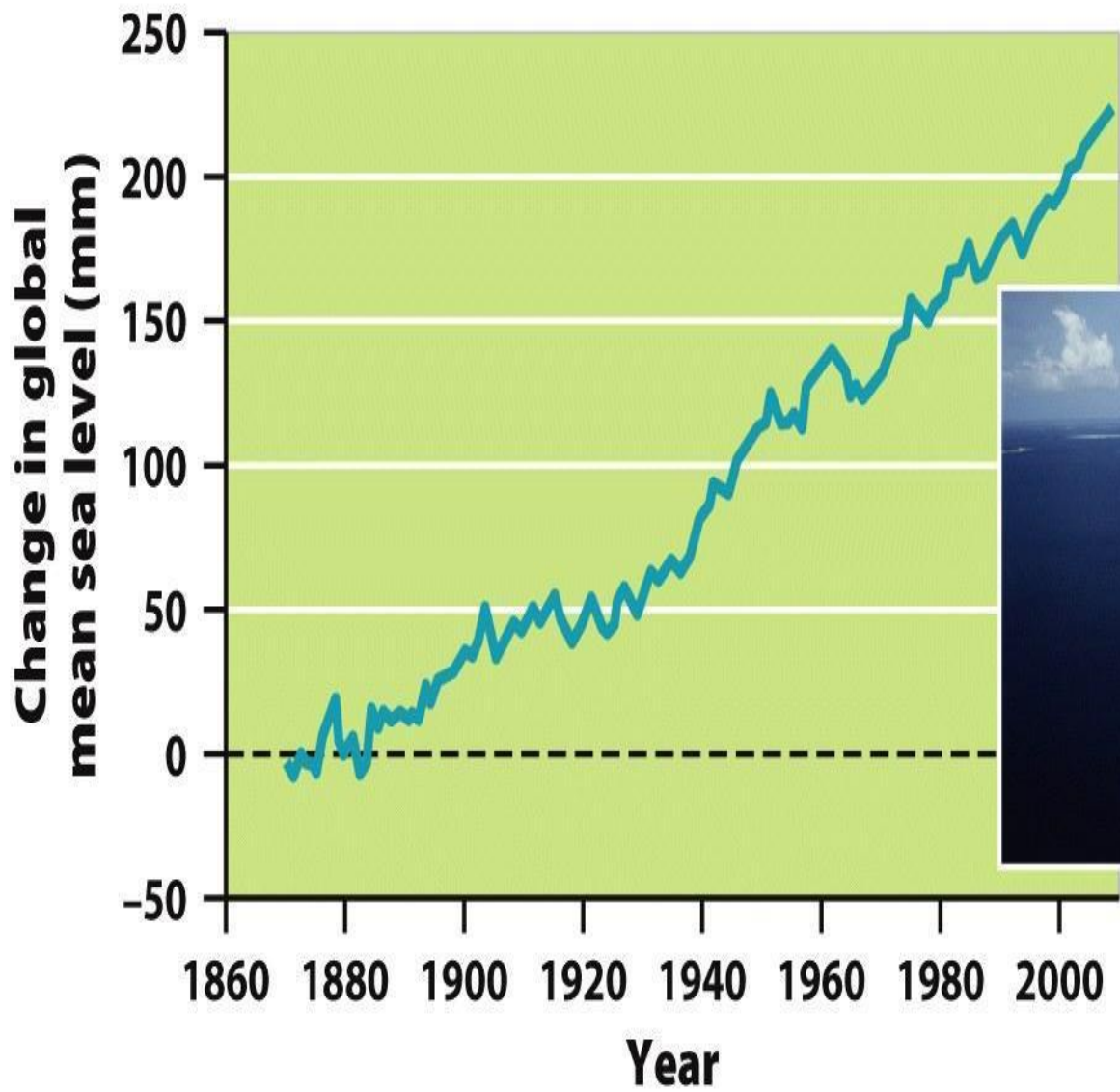
## Negative feedback system

Figure 19.18b  
Environmental Science  
© 2012 W. H. Freeman and Company

# Consequences to the Environment Because of Global Warming

- ❑ Melting of polar ice caps, Greenland and Antarctica
- ❑ Melting of many glaciers around the world (*could reduce seasonal water drinking supplies*)
- ❑ Melting of permafrost
- ❑ Rising of sea levels due to the melting of glaciers and ice sheets and as water warms it expands
- ❑ Heat waves
- ❑ Cold spells
- ❑ Change in precipitation patterns
- ❑ Increase in storm intensity
- ❑ Shift in ocean currents

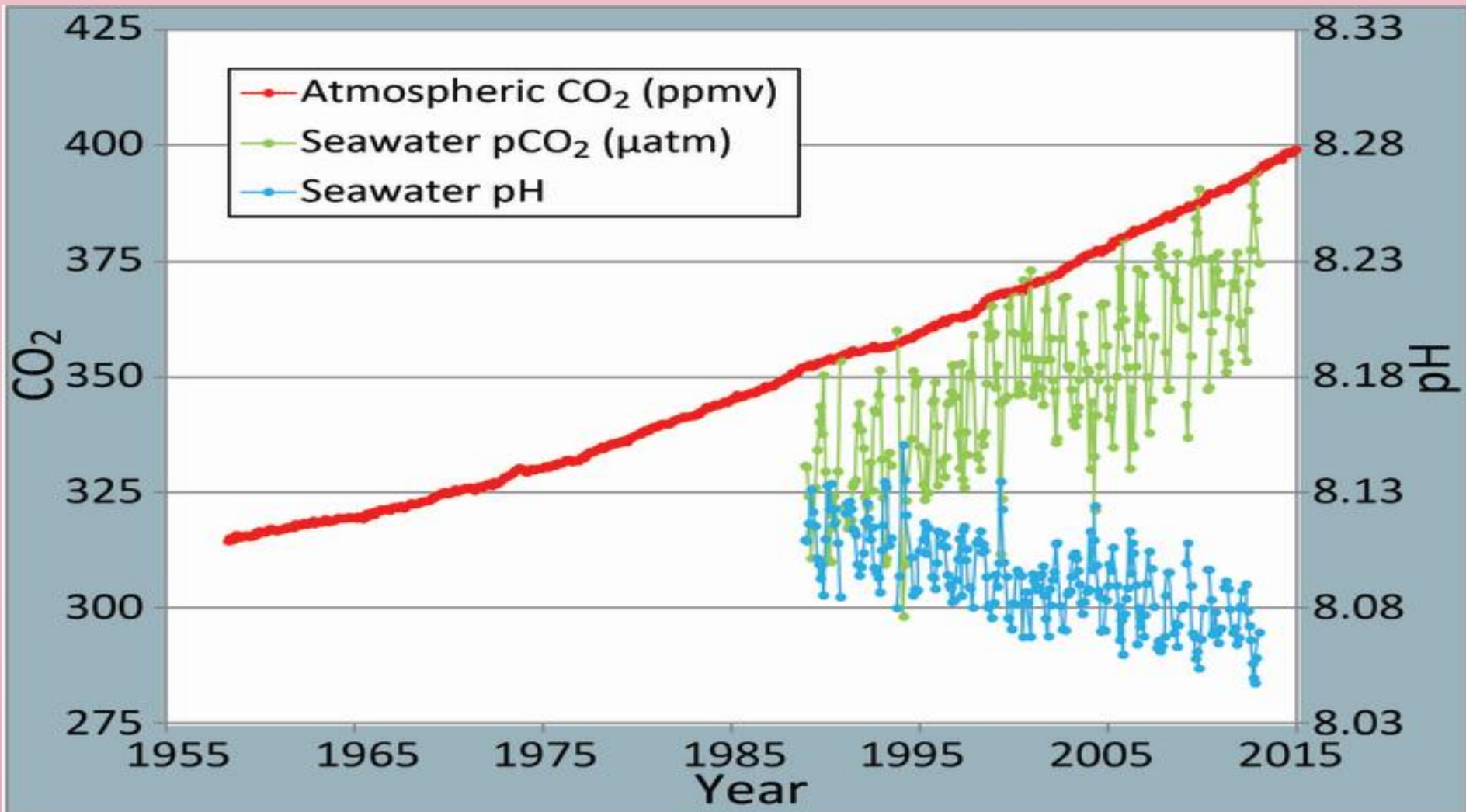




**Figure 19.20**

*Environmental Science*

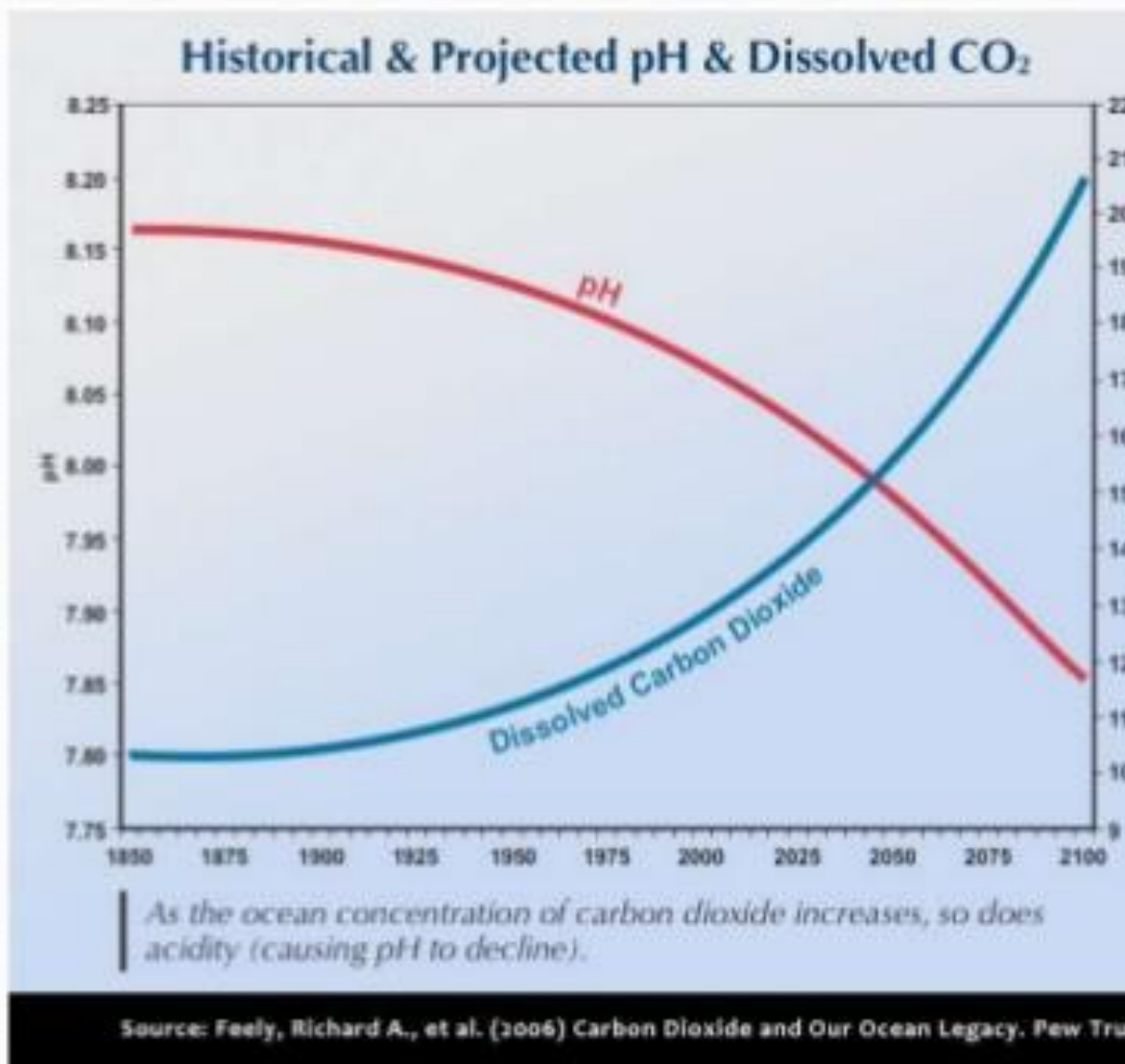
© 2012 W. H. Freeman and Company



This graph shows the correlation between rising levels of carbon dioxide (CO<sub>2</sub>) in the atmosphere at Mauna Loa with rising CO<sub>2</sub> levels in the nearby ocean at Station Aloha. As more CO<sub>2</sub> accumulates in the ocean, the pH of the ocean decreases. (modified after R. A. Feely, Bulletin of the American Meteorological Society, July 2008).

# What happens when CO<sub>2</sub> goes into the ocean?

- ~30% of the CO<sub>2</sub> emitted into the atmosphere dissolves into the ocean
- When CO<sub>2</sub> dissolves into water, it forms an acid
- The pH of the ocean has dropped 30% since the industrial revolution



# Consequences to Living Organisms

- Wild plants and animals can be affected. The growing season for plants has changed and animals have the potential to be harmed if they can't move to better climates.
  - *Beef cattle produce the most greenhouse gas emitted of all livestock*
- Humans may have to relocate, some diseases like those carried by mosquitoes could increase and there could be economic consequences.

# The Controversy of Climate Change

- The fundamental basis of climate change—that **greenhouse gas concentrations are increasing** and that this will lead to global warming is not in dispute among the vast majority of scientists.
- What is **unclear is how much world temperatures will increase for a given change in greenhouse gases**, because that depends on the different feedback loops.

**TABLE 19.2****The 2007 assessment of global change by the Intergovernmental Panel on Climate Change (IPCC)**

The scientists considered the likelihood that specific changes have occurred, the likelihood that humans contributed to the change, and the likelihood that current trends will continue.

Definitions: More likely than not = more than 50% certain; Likely = more than 60% certain; Very likely = more than 90% certain; Virtually certain = more than 99% certain.

Phenomenon and direction of trend	Likelihood that trend occurred in late 20th century (typically post-1960)	Likelihood of a human contribution to observed trend	Likelihood of future trends based on projections for 21st century from <i>Special Report on Emissions Scenarios</i>
Warmer and fewer cold days and nights over most land areas	Very likely	Likely	Virtually certain
Warmer and more frequent hot days and nights over most land areas	Very likely	Likely (nights)	Virtually certain
Warm spells/heat waves. Frequency increases over most land areas	Likely	More likely than not	Very likely
Heavy precipitation events. Frequency (or proportion of total rainfall from heavy falls) increases over most areas	Likely	More likely than not	Very likely
Area affected by droughts increases	Likely in many regions since 1970s	More likely than not	Likely
Intense tropical cyclone activity increases	Likely in some regions since 1970	More likely than not	Likely
Increased incidence of extreme high sea level (excludes tsunamis)	Likely	More likely than not	Likely

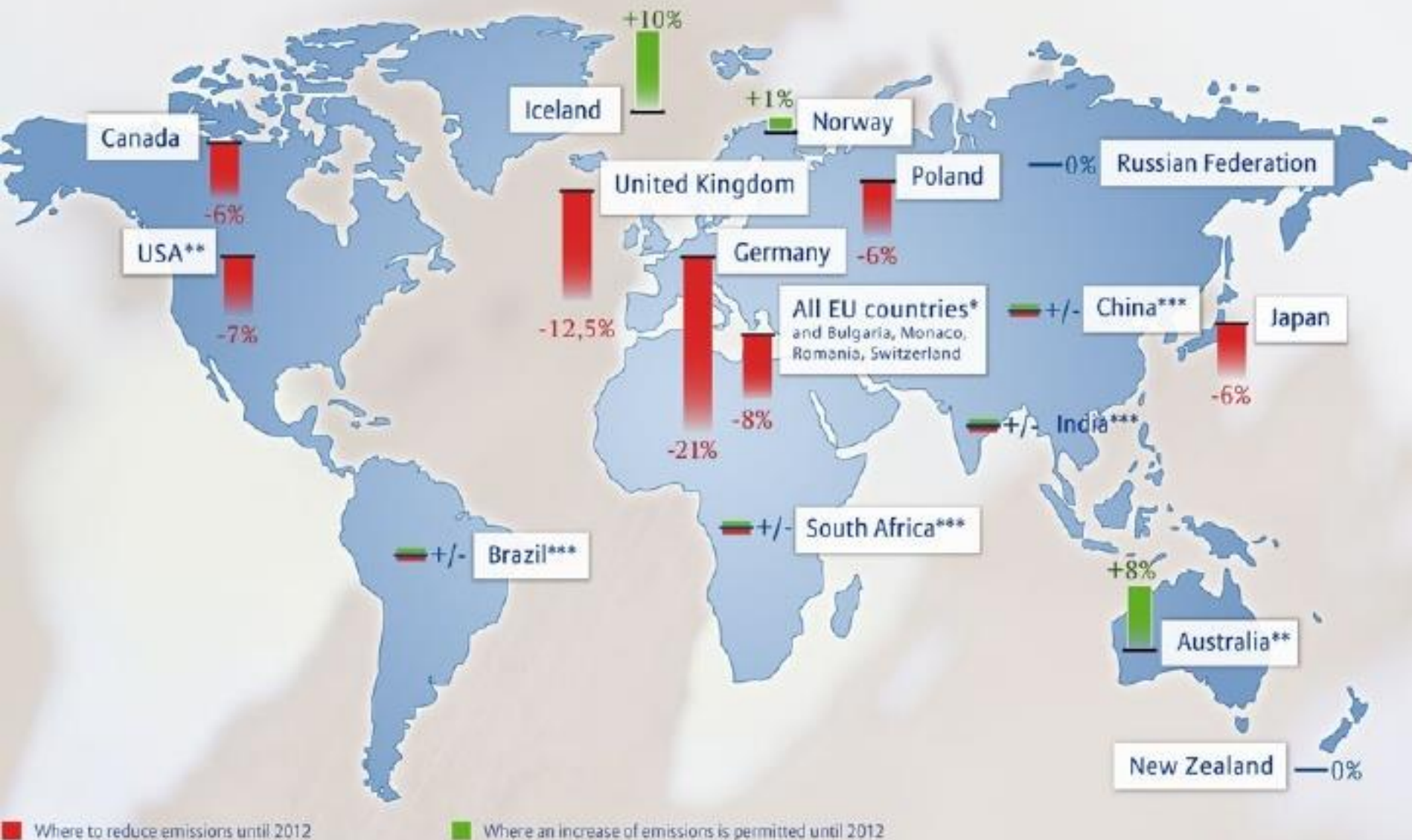
**Table 19.2***Environmental Science*

© 2012 W. H. Freeman and Company

# The Kyoto Protocol

- In 1997, representatives of the nations of the world went to Kyoto, Japan to discuss how best to control the emissions contributing to global warming.
- The **agreement** was that **emissions of greenhouse gases from all industrialized countries will be reduced to 5.2% below their 1990 levels by 2012.**
- *Developing nations did not have emission limits imposed by the protocol.*

## Emission Targets for Selected Countries (Kyoto Protocol)



\* The EU countries have redistributed their reductions commitments in a so-called burden-sharing commitment.

\*\* The USA and Australia have not ratified the protocol.

\*\*\* No restrictions under the Kyoto Protocol



# Problems

- Some delayed in signing up to Kyoto such as Russia who signed in 2004
- Some still have not. Australia the world's 2<sup>nd</sup> largest polluter per capita.
- USA initially signed but then withdrew in 2001 following GW Bush's election (USA emit 25% of world emissions)

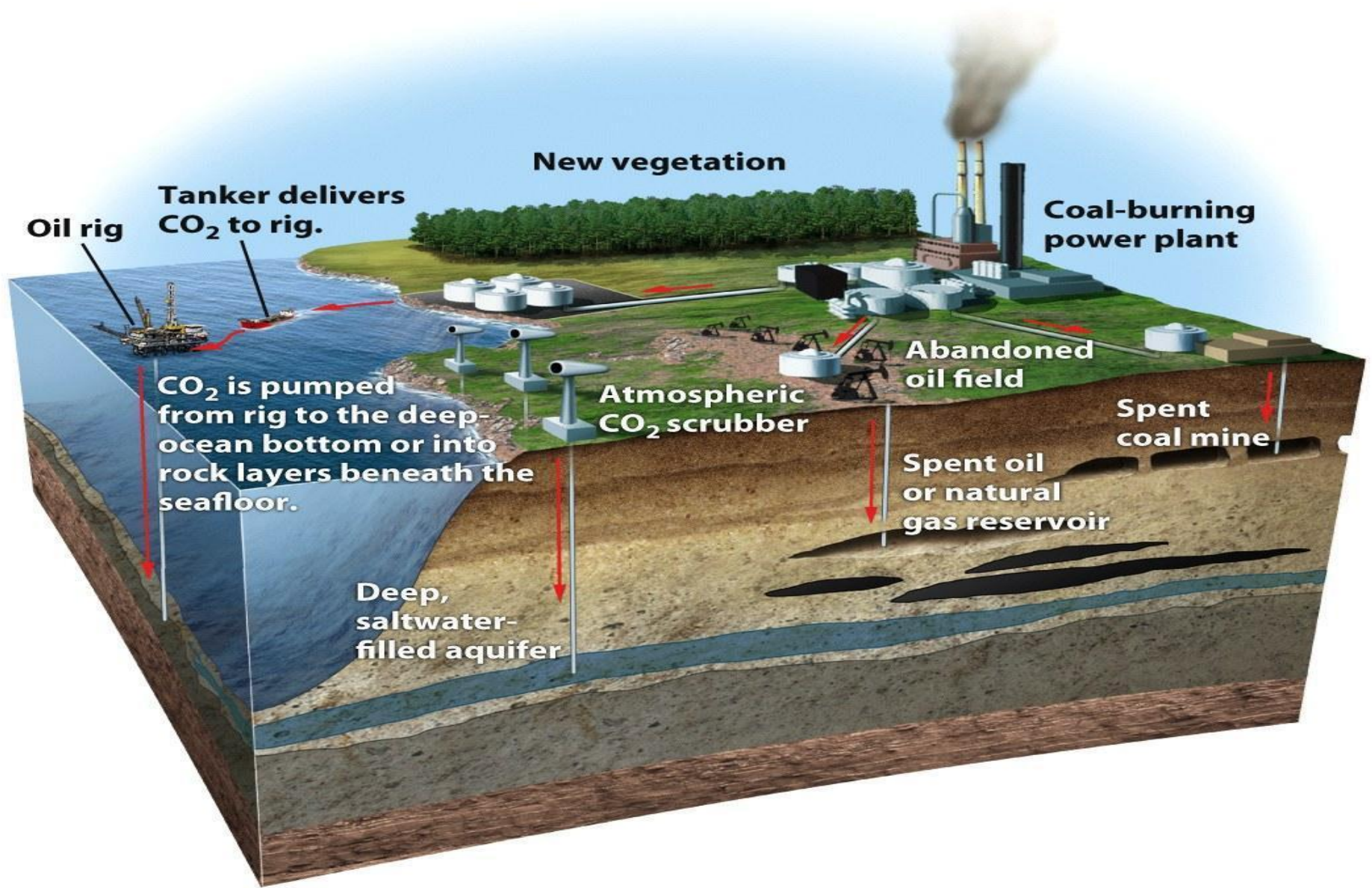
**The United States hasn't become part of the agreement because it considers a problem the fact that several major developing nations, including India and China, are not required to reduce emissions under the agreement.**

# UK Government Response

- Set targets to reduce emissions by 30% by 2020 and 60% by 2050
- Pass laws on carbon reduction targets
- Invest in green technology creating 100,000 new jobs
- Create a \$20 billion World Bank Fund to help poorer countries

# Carbon Sequestration

- An approach involving taking emitted CO<sub>2</sub> out of the atmosphere.
- Some methods include storing carbon in agricultural soils or retiring agricultural land and allowing it to become pasture or forest.
- Researchers are looking at cost-effective ways of capturing CO<sub>2</sub> from the air, from coal-burning power stations, and from other emission sources.
- This captured CO<sub>2</sub> would be compressed and pumped into abandoned oil wells or the deep ocean.



**To help reduce the amount of carbon dioxide that ends up in the atmosphere, efforts are being made to capture carbon at its source and either to convert it into plant biomass and soil carbon or to pump it down into the ground or the deep sea.**